

Aerosol Files in HITRAN

Some files in the ascii folder are carried over from previous HITRAN editions. The older files are listed first, followed by the newer files of aerosol indices of refraction. The * symbol in the file name indicates that multiple files (e.g. at several temperatures or weight percent concentrations) are associated with the measurements. If you have any questions, please contact Steven Massie at the Laboratory for Atmospheric and Space Physics (LASP), University of Colorado, Boulder, Colorado (1-303-735-6583, Steven.Massie@lasp.colorado.edu).

<u>File names</u>	<u>Substance</u>
downing_williams_water.dat	Water
kou_water_ice.dat	Water and Ice
afcr11987_shettle.dat	Water, Ice, Sodium Chloride, Sea Salt, Water Soluble Aerosol, Ammonium Sulfate, Carbonaceous Aerosol, Volcanic Dust, Sulfuric Acid, Meteoric Dust, Quartz, Hematite, Sand, and Dust-like Aerosol
palmer_williams_h2so4.dat	Sulfuric Acid Solutions
remsberg_h2so4_hno3.dat	Sulfuric Acid and Nitric Acid Solutions
steele_hamill_h2so4w.dat	Sulfuric Acid Solutions
timmermans_h2so4rho.dat	Sulfuric Acid Solutions
toon_psc.dat	H ₂ O -Ice, Amorphous Nitric Acid Solutions, and Nitric Acid Hydrates
norman_hno3_h2o.dat	Aqueous HNO ₃ / H ₂ O
tisdale_h2so4.dat	Sulfuric Acid Solutions
h2so4T*.biermann	Sulfuric Acid Solutions at 12 Temperatures
h2so4T*.niedziela	Sulfuric Acid Solutions at 8 Temperatures
querry_tyler_hno3_h2o.dat	Nitric Acid Solutions
hno3T*.biermann	Nitric Acid Solutions at 7 Temperatures
niedziela_nad.dat	Nitric Acid Dyhydrate (NAD) at 3 Temperatures
richwine_nat.dat	Nitric Acid Trihydrate (NAT)
ice*.clapp	Water Ice at 9 Temperatures
warren_ice.dat	Ice
wagner_supercooled.dat	Supercooled water
myhreh2so4*.dat	H ₂ SO ₄ /H ₂ O Solutions
myhrebin*hno3.dat	HNO ₃ /H ₂ O Solutions
myhretern*h2so4*hno3.dat	H ₂ SO ₄ /HNO ₃ /H ₂ O Solutions
sutherland_khanna_biomass.dat	Organic-based Nonvolatile Aerosols
magi_biomass_fire.dat	Vegetation-fire indices from field measurements
stagg_carbon.dat	Carbon flame at several temperatures
chang_flame_soot.dat	Flame soot
querry_diesel_soot.dat	Diesel soot
alexander_brown_carbon.dat	Brown carbon

organic_acids subdirectory	Organic acids (oxalic, malonic, succinic, pinonic, pyruvic, phthalic)
hasenkopf_organic_haze.dat	Organic haze
zarzana_soa.dat	Secondary organic aerosol proxy
querry_minerals.dat	Minerals (clay, illite, kaolin, montmorillonite)
toon_mont_granite.dat	Minerals (granite, montmorillonite)
wagner_sahara_dust.dat	Sahara dust at several hematite concentrations
sinyuk_saharan_dust.dat	Sahara dust indices from satellite data
grainger_volcanic_ash.dat	Volcanic ash
henning_mutschke_sio2_*.dat	Low temperature SiO ₂ (10K – 300 K)
zeidler_sio2_*k*.dat	High temperature SiO ₂ (300 K – 928 K; ab, c)
begemann_al2o3_*.dat	Al ₂ O ₃ (porous, compact)
henning_feo.dat	FeO
posch_catio3.dat	CaTiO ₃ (perovskite)
triaud_fe2o3.dat	Fe ₂ O ₃
jena_fe2sio4.dat	Fe ₂ SiO ₄ (fayalite)
fabian_fe2sio4_*.dat	Fe ₂ SiO ₄ (fayalite; x,y,z)
fabian_mgal2o4_*.dat	MgAl ₂ O ₄ (natural and annealed)
jager_mg2sio4.dat	Mg ₂ SiO ₄
jager_mgsio3.dat	MgSiO ₃
zeidler_tio2_rutile*.dat	TiO ₂ (rutile; ordinary, extraordinary)
zeidler_tio2_anatase_*.dat	TiO ₂ (anatase; ab, c)
posch_tio2_brookite_*.dat	TiO ₂ (brookite; x,y,z)
querry_kcl.dat	KCl
querry_zns.dat	ZnS
khare_tholins.dat	Titan Tholins
ramirez_titan_aerosol.dat	Titan Aerosol

File: downing_williams_water.dat

Data: Real and imaginary indices of refraction for water at 27 C between 10 and 5000 cm⁻¹.

Reference: H. D. Downing and D. Williams, "Optical constants of water in the infrared," *J.Geophys.Res.* **80**, 1656-1661 (1975).

File: kou_water_ice.dat

Data: Imaginary indices of refraction for water and ice in the 0.67 to 2.5 micron range.

Reference: L. Kou, D. Labrie, and P. Chylek, "Refractive indices of water and ice in the 0.65 to 2.5 micron range," *Appl.Opt.* **32**, 3531-3540 (1993).

File: afcrl1987_shettle.dat

Data: Real and imaginary indices of refraction for water, ice, sodium chloride, sea salt, water soluble aerosol, ammonium sulfate, carbonaceous aerosol, volcanic dust, sulfuric acid, meteoric dust, quartz, hematite, sand, and dust-like aerosol. The tabulations start at 0.2 micrometers and extend out to 40 micrometers (or to longer wavelengths).

These indices were used to generate the aerosol models which are incorporated into the LOWTRAN, MODTRAN, and FASCODE computer codes.

Reference: The data were tabulated by E.P. Shettle of the Naval Research Laboratory.

File: palmer_williams_h2so4.dat

Data: Real and imaginary indices of refraction of sulfuric acid solutions at 25, 38, 50, 75, 84.5, and 95.6% H₂SO₄, by weight.

Reference: K.F. Palmer and D. Williams, "Optical constants of sulfuric acid; Application to the clouds of Venus?" *Appl. Opt.* **14**, 208-219 (1975).

File: remsberg_h2so4_hno3.dat

Data: Real and imaginary indices of refraction of sulfuric acid solutions at 75 and 90% H₂SO₄, by weight, plus the standard deviations of the measurements. Real and imaginary indices of refraction of nitric acid solutions at 68% HNO₃, by weight, plus the standard deviations of the measurements.

Reference: E.E. Remsberg, D. Lavery, and B. Crawford, "Optical constants for sulfuric and nitric acids," *J.Chem.Engin.Data* **19**, 263-255 (1974).

File: steele_hamill_h2so4w.dat

Data: Theoretical equilibrium composition (weight percentage of H₂SO₄) of sulfuric acid

solutions, given as a function of temperature and H₂O vapour pressure.

Reference: H.M. Steele and P. Hamill, "Effects of temperature and humidity on the growth and optical properties of sulphuric acid-water droplets in the stratosphere," *J.Aerosol Sci.* **12**, 517-528 (1981).

File: timmermans_h2so4rho.dat

Data: Sulfuric acid density (gm/cm³) values for solutions between 0 and 100% H₂SO₄ (by weight) for temperatures between 0 and 60 C.

Reference: J. Timmermans, "The physico-chemical constants of binary systems in concentrated solutions," Interscience, New York, (1960).

The data cited here (from pages 561-562 of Timmermans) are from Domke and Bein (1905).

File: toon_psc.dat

Data: Real and imaginary indices of refraction of H₂O-ice, amorphous nitric acid solutions, and nitric acid hydrates.

Reference: O.B. Toon, M.A. Tolbert, B.G. Koehler, A.M. Middlebrook, and J. Jordan, "The infrared optical constants of H₂O-ice, amorphous acid solutions, and nitric acid hydrates," *J.Geophys.Res.* **99**, 25631-25654 (1994).

File: norman_hno3_h2o.dat

Data: Real and imaginary indices of aqueous HNO₃/H₂O at 220 K from 754 to 4700 cm⁻¹ for 35, 45, 54, 63, and 70% HNO₃ by weight.

Reference: M.L. Norman, J. Qian, R.E. Miller, and D.R. Worsnop, "Infrared complex refractive indices of supercooled liquid HNO₃/ H₂O aerosols," *J.Geophys.Res.* **104**, 30571-30584 (1999).

Email contact person: R.E. Miller (remiller@unc.edu)

File: **tisdale_h2so4.dat**

Data: Real and imaginary indices of liquid H₂SO₄/ H₂O at 215 K from 499 to 6996 cm⁻¹ as a function of the H₂SO₄ concentration by weight.

Reference: R.T. Tisdale, D.L. Glandorf, M.A. Tolbert, and O.B. Toon, "Infrared optical constants of low-temperature H₂SO₄ solutions representative of stratospheric sulfate aerosols," *J.Geophys.Res.* **103**, 25353-25370 (1998).

Email contact person: M. Tolbert (tolbert@spot.colorado.edu)

Files: **h2so4T*.biermann**

Data: Real indices of liquid H₂SO₄/ H₂O from 0 to 16382 cm⁻¹ and imaginary indices from 432 to 5028 cm⁻¹ at 12 temperatures (213 to 293K) as a function of the H₂SO₄ concentration by weight.

Reference: U.M. Biermann, B.P. Luo, and Th. Peter, "Absorption Spectra and Optical Constants of Binary and Ternary Solutions of H₂SO₄, HNO₃, and H₂O in the Mid Infrared at Atmospheric Temperatures," *J.Phys.Chem. A* **104**, 783-793 (2000).

Email contact person: B.P. Luo (luo@mpch-mainz.mog.de)

Files: **h2so4T*.niedziela**

Data: Real and imaginary indices of liquid H₂SO₄/ H₂O at 8 temperatures (200 to 300 K) from 825 to 4700 cm⁻¹.

Reference: R.F. Niedziela, M.L. Norman, C.L. deForest, R.E. Miller, and D.R. Worsnop, "A Temperature and Composition-Dependent Study of H₂SO₄ Aerosol Optical Constants Using Fourier Transform and Tunable Diode Laser Infrared Spectroscopy," *J.Phys.Chem. A* **103**, 8030-8040 (1999).

Email contact person: R.E. Miller (remiller@unc.edu)

File: **querry_tyler_hno3_h2o.dat**

Data: Real and imaginary indices of liquid HNO₃/ at room temperature from 250 to 4987 cm⁻¹ as a function of the HNO₃ concentration by weight. The concentrations of 3.1, 6.1, 11.8, 22.3, 40.3, and 70% HNO₃ correspond to the molar (M) concentrations of 0.5, 1.0, 2.0, 4.0, 8.0, and 15.7 cited in the Querry and Tyler paper.

Reference: M.R. Querry and I.L. Tyler, "Reflectance and complex refractive indices in the infrared of aqueous solutions of nitric acid," *J.Chem.Phys.* **72**, 2495-2499 (1980).

Files: **hno3T*.biermann**

Data: Real indices of liquid HNO₃/ H₂O from 0 to 16382 cm⁻¹ and imaginary indices from 432 to 5028 cm⁻¹ at 7 temperatures (213 to 293K) as a function of the HNO₃ concentration by weight.

Reference: U.M. Biermann, B.P. Luo, and Th. Peter, "Absorption Spectra and Optical Constants of Binary and Ternary Solutions of H₂SO₄, HNO₃, and H₂O in the Mid Infrared at Atmospheric Temperatures," *J.Phys.Chem. A* **104**, 783-793 (2000).

Email contact person: B.P. Luo (luo@mpch-mainz.mog.de)

Files: **niedziela_nad.dat**

Data: Real and imaginary indices of nitric acid dihydrate (NAD) at 3 temperatures (160 to 190 K) from 700 to 4700 cm⁻¹.

Reference: R.F. Niedziela, R.E. Miller, and D.R. Worsnop, "Temperature and Frequency- Dependent Optical Constants for Nitric Acid Dihydrate from Aerosol Spectroscopy," *J.Phys.Chem. A* **102**, 6477-6484 (1998).

Email contact person: R.E. Miller (remiller@unc.edu)

File: **richwine_nat.dat**

Data: Real and imaginary refractive indices of nitric acid trihydrate (NAT) at 160 K from 711 to 4004 cm⁻¹.

Reference: L.J. Richwine, M.L. Clapp, R.E. Miller, and D.R. Worsnop, "Complex refractive indices in the infrared of nitric acid trihydrate aerosols," *Geophy.Res.Lett.* **22**, 2625-2628 (1995).

Email contact person: R.E. Miller (remiller@unc.edu)

Files: **ice*.clapp**

Data: Real and imaginary indices of water ice at 9 temperatures (130 to 210 K) from 800 to 4004 cm⁻¹.

Reference: M.L. Clapp, R.E. Miller, and D.R. Worsnop, "Frequency-Dependent Optical Constants of Water Ice Obtained Directly from Aerosol Extinction Spectra," *J.Phys.Chem.* **99**, 6317-6326 (1995).

Email contact person: R.E. Miller (remiller@unc.edu)

File: **warren_ice.dat**

Data: Real and Imaginary indices of Ice Ih at 266 K from 44 nm to 2 m.

Reference: S.G. Warren and R.E. Brandt, "Optical constants of ice from the ultraviolet to the microwave: A revised compilation." *J.Geophys.Res.* **113**, D14220 (2008).

Email contact person: Stephen G. Warren (sgw@atmos.washington.edu)

File: **wagner_supercooled.dat**

Data: Real and imaginary indices of supercooled water at 238, 252, 258, and 269 K from 1101 to 4503 cm⁻¹

Reference: R. Wagner, S. Benz, O. Muhler, H. Saathoff, M. Schnaiter, and U. Schurath, "Mid- Infrared Extinction Spectra and Optical Constants of Supercooled Water Droplets," *J.Phys.Chem.* **109**, 7099-7112 (2005).

Email contact person: Robert Wagner (Robert.Wagner@imk.fzk.de)

Files: **myhreh2so4*.dat**

Data: Real and imaginary indices of H₂SO₄/H₂O droplets at 0.81, 0.81, 0.81, 0.76, 0.76 weight percent H₂SO₄ at 298, 273, 267, 298, 273 K from 1.3 to 25 μm.

Reference: C.E. Lund Myhre, D.H. Christensen, F.M. Nicolaisen, and C.J. Nielsen, "Spectroscopic Study of Aqueous H₂SO₄ at Different Temperatures and Compositions: Variations in Dissociation and Optical Properties," *J.Phys.Chem.* **107**, 1979-1991 (2005).

Email contact person: C.E. Lund Myhre (e.c.lund@iakh.uio.no.)

Files: **myhrebin*hno3.dat**

Data: Real and imaginary indices of binary HNO₃/H₂O at 223, 233, 243, 253, 273 and 293 K from 450 to 6500 cm⁻¹ 30% HNO₃, 70% H₂O from 1.5 to 22 μm.

Reference: C.E. Lund Myhre, H. Grothe, A.A. Gola, and C.J. Nielsen, "Optical Constants of HNO₃/H₂O and H₂SO₄/HNO₃/H₂O at Low Temperatures in the Infrared Region," *J.Phys.Chem.* **109**, 7166-7171 (2005).

Email contact person: C.J. Nielsen (c.j.nielsen@kjemi.uio.no.)

Files: **myhretern*h2so4*hno3.dat**

Data: Real and imaginary indices of ternary droplets at 203, 213, 223, 253, 273 and 293 K from 450 to 6500 cm⁻¹ 23% HNO₃, 21% H₂SO₄, 56% H₂O from 1.5 to 22 μm.

Reference: C.E. Lund Myhre, H. Grothe, A.A. Gola, and C.J. Nielsen, "Optical Constants of HNO₃/H₂O and H₂SO₄/HNO₃/H₂O at Low Temperatures in the Infrared Region," *J.Phys.Chem.* **109**, 7166-7171 (2005).

Email contact person: C.J. Nielsen (c.j.nielsen@kjemi.uio.no.)

File: **sutherland_khanna_biomass.dat**

Data: Real and imaginary refractive indices of organic-based nonvolatile aerosols produced by burning vegetation from 525 to 5000 cm⁻¹. The mixed weed sample indices of Table 2 from the paper by Sutherland and Khanna are tabulated here.

Reference: R.A. Sutherland and R.K. Khanna, "Optical Properties of Organic-based Aerosols Produced by Burning Vegetation," *Aerosol Science and Technology* **14**, 331-342 (1991).

File: **magi_biomass_fire.dat**

Data: Real and imaginary indices of biomass burning aerosol retrieved from SAFARI field campaign data, from 354 - 1557 nm

Reference: B. J. Magi, Q. Fu, and Redemann, A methodology to retrieve self-consistent aerosol properties using common aircraft measurements *J. Geophys. Res.*, **112**, D24S12, doi:10.1029/2006JD008312, 2007

File: **stagg_carbon.dat**

Data: Real and imaginary indices of refraction of pyrolytic graphite, amorphous carbon, propane and flame soot 400 to 700 nm

Reference: B. J. Stagg and T. T. Charalampopoulos, Refraction Indices of Pyrolytic Graphite, Amorphous Carbon, and Flame Soot in the Temperature Range 25 to 600C, *Combustion and Flame*, **94**:381-396, 1993.

File: **chang_flame_soot.dat**

Data: Real and imaginary indices of refraction of flame soot from 0.20 to 38.4 microns at three heights (in mm) above the flame

Reference: H. Chang and T. T. Charalampopoulos, Determination of the Wavelength Dependence of Refractive Indices of Flame Soot

Proc. R. Soc. London A 1990, **430**, 577-591, doi:10.198/rspa.1990.0107

File: **querry_diesel_soot.dat**

Data: Real and imaginary indices of refraction of Diesel Soot.

Reference: Marvin R. Querry, Optical Constants of Minerals and Other Materials From The Millimeter To The Ultraviolet CRDEC-CR-88009, November 1987.

File: **alexander_brown_carbon.dat**

Data: Real and imaginary indices of brown carbon spheres measured during the Asian Pacific Regional Aerosol Characterization Experiment (ACE-Asia). Complex dielectric constants are also tabulated.

Reference: Duncan. T. Alexander, Peter A. Crozier, James R. Anderson, Brown Carbon Spheres in East Asian Outflow and Their Optical Properties, *Science*, **321**, 8, August 2008.

Email contact person: Peter Crozier (crozier@asu.edu).

File: **organic_acids_subdirectory**

Data: Imaginary indices of Organic acids (oxalic, malonic, succinic, pinonic, pyruvic, phthalic) at 293K from 0.25 to 1.1 microns.

Reference: C.E. Lund Myhre and C.J. Nielsen, Optical properties in the UV and visible spectral region of organic acids relevant to tropospheric aerosols, *Atmos. Chem. Phys.*, **4**, 1759-1769, 2004.

Email contact person: C. J. Nielsen (claus.nielsen@kjemi.uio.no)

File: **hasenkopf_organic_haze.dat**

Data: Real and imaginary indices at 532 nm of organic haze for Titan and early Earth, and other determinations.

Reference: C. Hasenkopf, M. Beaver, M. Trainer, H. Dewit, M. Freedman, O. Toon, C. McKay, M. Tolbert, Optical properties of Titan and early earth haze laboratory analogs in the mid-visible, *Icarus*, **207**, 903-913, 2010.

Email contact person: Christa Hasenkopf (Christa.Hasenkopf@colorado.edu)

File: **zarzana_soa.dat**

Data: Real and imaginary indices at 532 nm of proxies of secondary organic aerosol, and other determinations of HULIS indices

Reference: Kyle J. Zarzana, David O. DeHaan, Miriam A. Freedman, Christa A. Hasenkopf, and Margaret A. Tolbert. Optical Properties of the Products of alpha-Dicarbonyl and Amine Reactions in Simulated Cloud Droplets, *Env Sci Tech*, **46**, 4845-4851, 2012.

Email contact person: Kyle Zarzana (kyle.zarzana@colorado.edu)

File: quarry_minerals.dat

Data: Real and imaginary indices of refraction of materials from 2.5 to 50 microns. This listing focuses upon various minerals (clay, illite, kaolin, montmorillonite).

Reference: Marvin R. Querry, Optical Constants of Minerals and Other Materials From The Millimeter To The Ultraviolet CRDEC-CR-88009, November 1987.

Email contact person: Steven Massie (Steven.Massie@lasp.colorado.edu)

File: toon_mont_granite.dat

Data: Real and imaginary indices of Montmorillonite and Granite from 250 to 2000 cm⁻¹.

Reference: Toon, O. B., J. B. Pollack, C. Sagan, Physical Properties of the Particles Composing the Martian Dust Storm of 1971 – 1972, *Icarus*, **30**, 663-696, 1977

Email contact person: Brian Toon, btoon@lasp.colorado.edu

File: wagner_sahara_dust.dat

Data: Imaginary indices of Saharan dust samples for five locations with hematite between 1 and 3 % by volume for 305 - 955 nm.

Reference: R. Wagner, T. Ajtai, K. Kandler, K. Lieke, C. Linke, T. Muller, M. Schnaiter, and N. Vragel. Complex refractive indices of Saharan dust samples at visible and near UV wavelengths: a laboratory study. *Atmos. Chem. Phys.*, **12**, 2491-2512, 2012.

Email contact person: R. Wagner (robert.wagner2@kit.edu)

File: sinyuk_saharan_dust.dat

Data: Imaginary indices of Saharan dust derived from TOMS data for 350 - 650 nm. Also include Patterson (1977) data in file.

Reference: Alexander Sinyuk, Omar Torres, and Oleg Dubovik, Combined use of satellite and surface observations to infer the imaginary part of refractive index of Saharan dust, *Geophys. Res. Lett.*, **30**, no 2, 1081, doi:10.1029/2002GL016189, 2003

Email contact person: Steven Massie (Steven.Massie@lasp.colorado.edu)

File: grainger_volcanic_ash.dat

Data: Ash volcanic refractive indices from 1 to 20 microns.

Reference: Grainger, R. G., Peters, D. M., Thomas, G. E., Smith, A. J. A., Siddans, R., Carboni, E., and Dudhia, A.: Measuring Volcanic Plume and Ash Properties from Space, in: Remote Sensing of Volcanoes and Volcanic Processes: Integrating Observation and Modelling, edited by Pyle, D. and Mather, T., vol. 270 of Special Publications, Geological Society, London, 2012.

Email contact person: R. G. Grainger (R.Grainger@physics.ox.ac.uk)

File: henning_mutschke_sio2_*.dat

Data: Real and Imaginary Refractive Indices of amorphous SiO₂ at 10, 100, 200, 300 K.

Reference: Henning, T. and H. Mutschke, Low Temperature infrared properties of cosmic dust analogues, *Astronomy and Astrophysics*, **327**, p743-754, 1997

Email contact person: Thomas Henning (henning@mpia.de)

File: **zeidler_sio2_*k*.dat**

Data: Real and Imaginary Refractive Indices of crystalline SiO₂ at 300, 551, 738, 833 K.

Reference: Zeidler, S., Th. Posch, and H. Mutschke, Optical constants of refractory oxides at high temperatures Mid-infrared properties of corundum, spinel, and alpha-quartz, *Astronomy and Astrophysics*, **553**, A81, 2013.

Email contacts: Simon Zeidler (simon.zeidler@nao.ac.jp) and H. Mutschke (harald.mutschke@uni-jena.de)

File: **begemann_al2o3_*.dat**

Data: Real and Imaginary Refractive Indices of compact and porous Al₂O₃.

Reference: Begemann, B., J. Dorschner, Th. Henning, H. Mutschke, J. Gurtler, C. Kompe, and R. Nass, Aluminum oxide and the opacity of oxygen-rich circumstellar dust in the 12-17 micron range, *Astrophysical Journal*, **476**, 199-208, 1997.

Email contact person: H. Mutschke (harald.mutschke@uni-jena.de)

File: **henning_feo.dat**

Data: Real and Imaginary Refractive Indices of FeO.

Reference: Henning, Th., B. Begemann, H. Mutschke, J. Dorschner, Optical properties of oxide dust grains, *Astronomy and Astrophysics, Supplement Series*, **112**, 143-149, 1995.

Email contact person: Thomas Henning (henning@mpia.de)

File: **posch_catio3.dat**

Data: Real and Imaginary Refractive Indices of CaTiO₃ (Perovskite).

Reference: Posch, Th., F. Kerschbaum, D. Fabian, H. Mutschke, J. Dorschner, A. Tamanai, Th. Henning, Infrared Properties of Solid Titanium Oxides: Exploring Potential Primary Dust Condensates *Astrophysical Journal Supplement Series*, **149**, 437-445, 2003.

Email contact person: Thomas Posch (thomas.posch@univie.ac.at)

File: **triaud_fe2o3.dat**

Data: Real and Imaginary Refractive Indices of Fe₂O₃ E || a,b.

Reference: HMJ Triaud (Amaury.Triaud@obs.unige.ch) unpublished data stored at <http://www.astro.uni-jena.de/Laboratory/OCDB/mgfeoxides.html#C>

Email contact person: Amaury H.M.T Triaud (aht34@ast.cam.ac.uk)

File: **jena_fe2sio4.dat**

Data: Real and Imaginary Refractive Indices of Fe₂SiO₄ (Fayalite).

Reference: unpublished data stored at http://www.astro.uni-jena.de/Laboratory/OCDB/data/silicate/crystalline/fay_vis.txt

Email contact person: H. Mutschke (harald.mutschke@uni-jena.de)

File: **fabian_fe2sio4_*.dat**

Data: Real and Imaginary Refractive Indices of Fe₂SiO₄ (Fayalite).

Reference: Fabian, D., T. Henning, C. Jager, H. Mutschke, J. Dorschner and O. Wehrhan, Steps toward interstellar silicate mineralogy VI. Dependence of crystalline olivine IR spectra on iron content and particle shape, *Astronomy and Astrophysics*, **378**, 228-238, 2001.

Email contact person: H. Mutschke (harald.mutschke@uni-jena.de)

File: **fabian_mgal2o4_*.dat**

Data: Real and Imaginary Refractive Indices of Annealed MgAl_2O_4 .

Reference: Fabian, D., Th. Posch, H. Mutschke, F. Kerschbaum, and J. Dorschner, Infrared optical properties of spinels, A study of the carrier of the 13, 17 and 32 micron emission features observed in ISO-SWS spectra of oxygen-rich AGB stars, *Astronomy and Astrophysics*, **373**, 1125-1138, 2001.

Email contact person: H. Mutschke (harald.mutschke@uni-jena.de)

File: **jager_mg2sio4.dat**

Data: Real and Imaginary Refractive Indices of Mg_2SiO_4 .

Reference: Jager, C., J. Dorschner, H. Mutschke, Th. Posch, Th. Henning, Steps toward interstellar silicate mineralogy VII. Spectral properties and crystallization behaviour of magnesium silicates produced by the sol-gel method, *Astronomy and Astrophysics*, **408**, 193-204, 2003.

Email contact person: Cornelia Jaeger (cornelia.jaeger@uni-jena.de)

File: **jager_mgsio3.dat**

Data: Real and Imaginary Refractive Indices of MgSiO_3 .

Reference: Jager, C., J. Dorschner, H. Mutschke, Th. Posch, Th. Henning, Steps toward interstellar silicate mineralogy VII. Spectral properties and crystallization behaviour of magnesium silicates produced by the sol-gel method, *Astronomy and Astrophysics*, **408**, 193-204, 2003.

Email contact person: Cornelia Jaeger (cornelia.jaeger@uni-jena.de)

File: **zeidler_tio2_rutile*.dat**

Data: Real and Imaginary Refractive Indices of TiO_2 (Rutile; Ordinary and Extraordinary).

Reference: Zeidler, S., T. Posch, H. Mutschke, H. Richter, O. Wehrhan, Near-infrared absorption properties of oxygen-rich stardust analogs, The influence of coloring metal ions, *Astronomy and Astrophysics*, **526**, article 68, 1-10, 2011.

Email contacts: Simon Zeidler (simon.zeidler@nao.ac.jp) and H. Mutschke (harald.mutschke@uni-jena.de)

File: **zeidler_tio2_anatase_*.dat**

Data: Real and Imaginary Refractive Indices of TiO_2 (Anatase; ab, c).

Reference: Zeidler, S., T. Posch, H. Mutschke, H. Richter, O. Wehrhan, Near-infrared absorption properties of oxygen-rich stardust analogs, The influence of coloring metal ions, *Astronomy and Astrophysics*, **526**, article 68, 1-10, 2011.

Email contacts: Simon Zeidler (simon.zeidler@nao.ac.jp) and H. Mutschke (harald.mutschke@uni-jena.de)

File: **posch_tio2_brookite_*.dat**

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Email contact person: Thomas Posch (thomas.posch@univie.ac.at)

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Email contact person: Steven Massie (Steven.Massie@lasp.colorado.edu)

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Email contact person: Steven Massie (Steven.Massie@lasp.colorado.edu)

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