

The CU Airborne SOF Instrument: Developing Techniques to Quantify Wildfire Emissions

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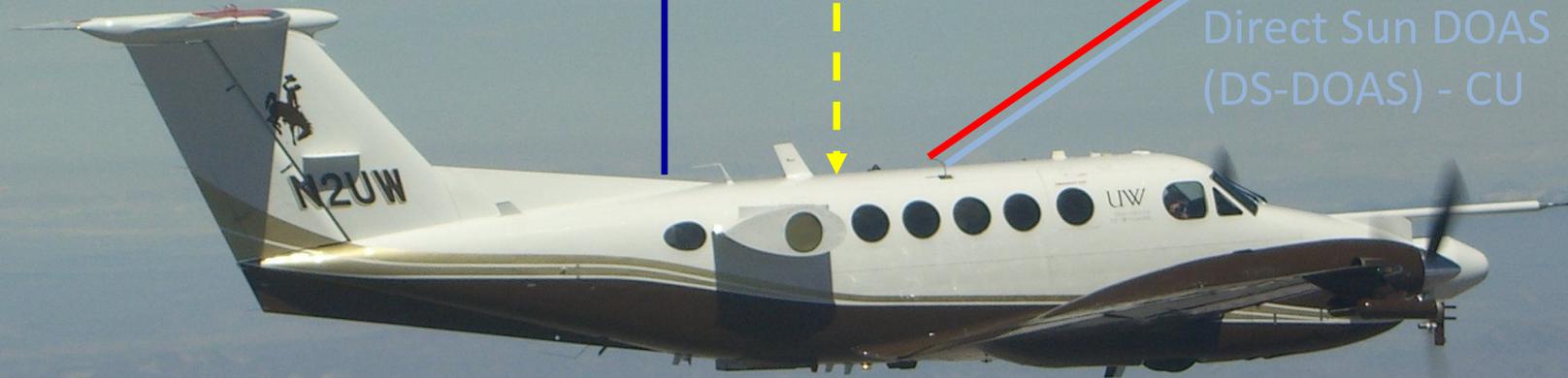
BB-FLUX payload

Zenith Sky DOAS
(ZS-DOAS) - CU

CU SOF

Wyoming Aerosol &
Cloud lidar (WCL v2)

Direct Sun DOAS
(DS-DOAS) - CU



in-situ sampling:

- CO, CO₂, H₂O
- O₃ (KIT)
- Aerosol size distribution
- Cloud size distributions

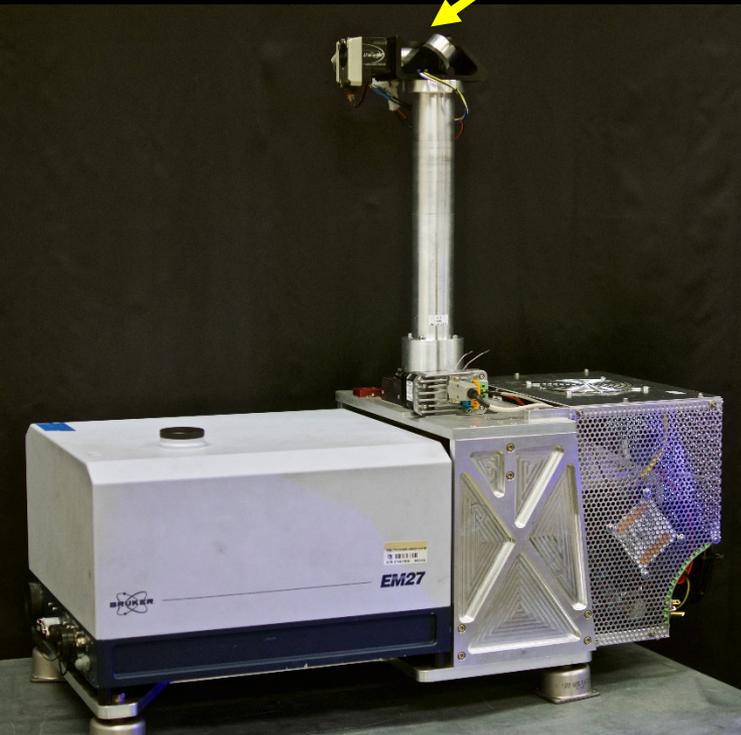
Auxiliary :

- Radiation (up- and downwelling)
- Video (forward & downward)

CU airborne SOF: emission flux & chemistry

University of Colorado airborne
Solar Occultation Flux

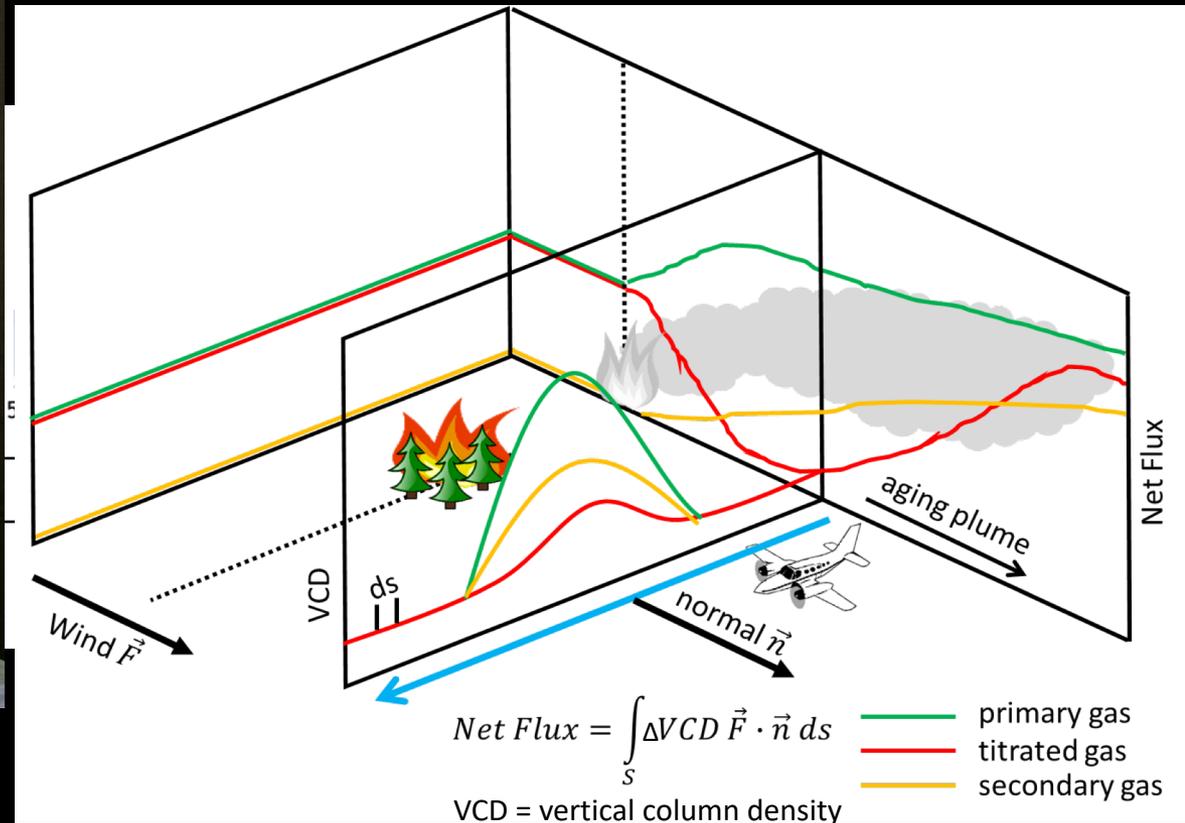
Direct sunlight
observation



Weight: ~38 kg

Size: 356 mm x 750 mm (W x L)

Column measurements are
independent of PBLH



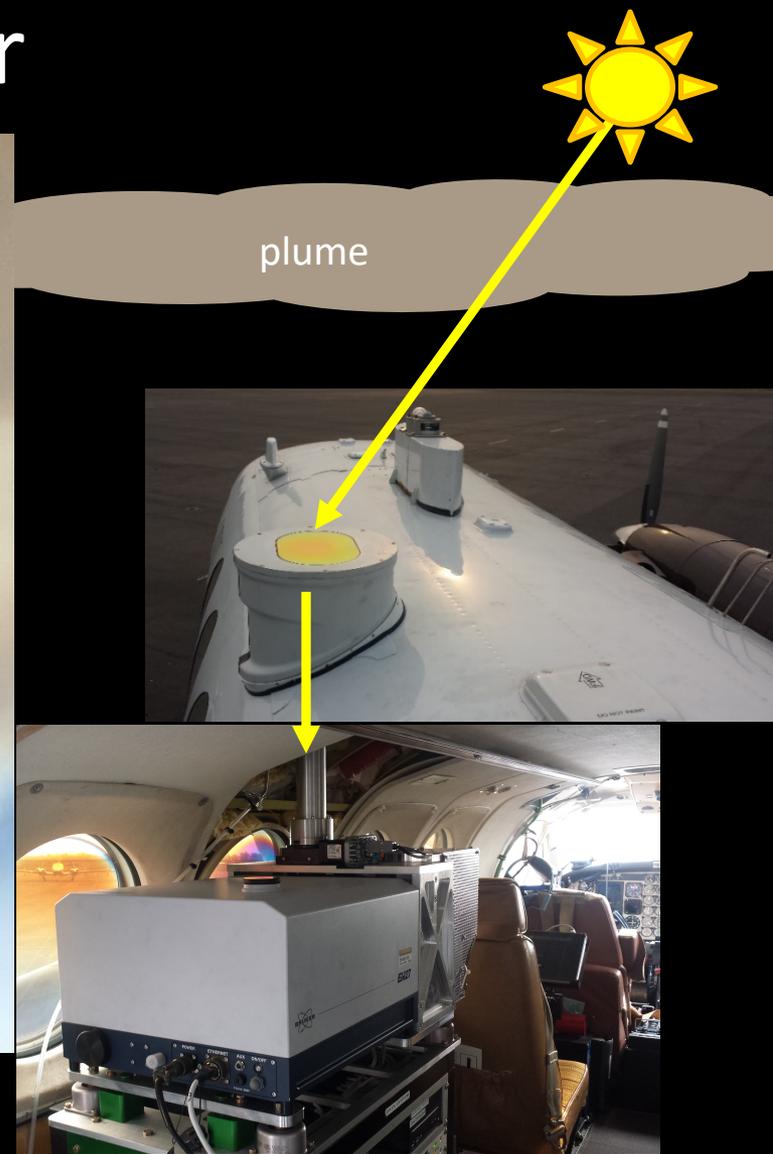
Hypothesis: It is possible to quantify trace
gas fluxes at the scale of wildfires.

Digital mobile solar tracker



Access to Euler Angles of the sun
Active control to correct for vehicle motion
→ Less time adjusting & more time measuring

Works under optically thick plumes & partly cloudy skies



Quantifying impacts on the atmosphere

Need to know what and how much is emitted from wildfires

Bottom up method: $E = A \cdot F \cdot \beta \cdot EF$

Top down method: $E = FRE \cdot c \cdot EF$

E = emission (g)

A = area burned (km²)

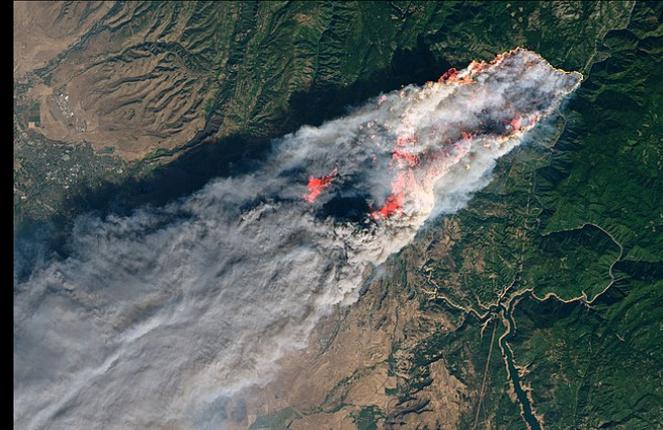
F = fuel load (kg/km²)

β = combustion completeness (0-1)

EF = emission factor (g/kg)

FRE = fire radiative energy (MJ)

c = conversion factor (kg/MJ)



San Francisco Chronicle

Camp Fire's climate toll: Greenhouse gases equal about a week of California auto emissions



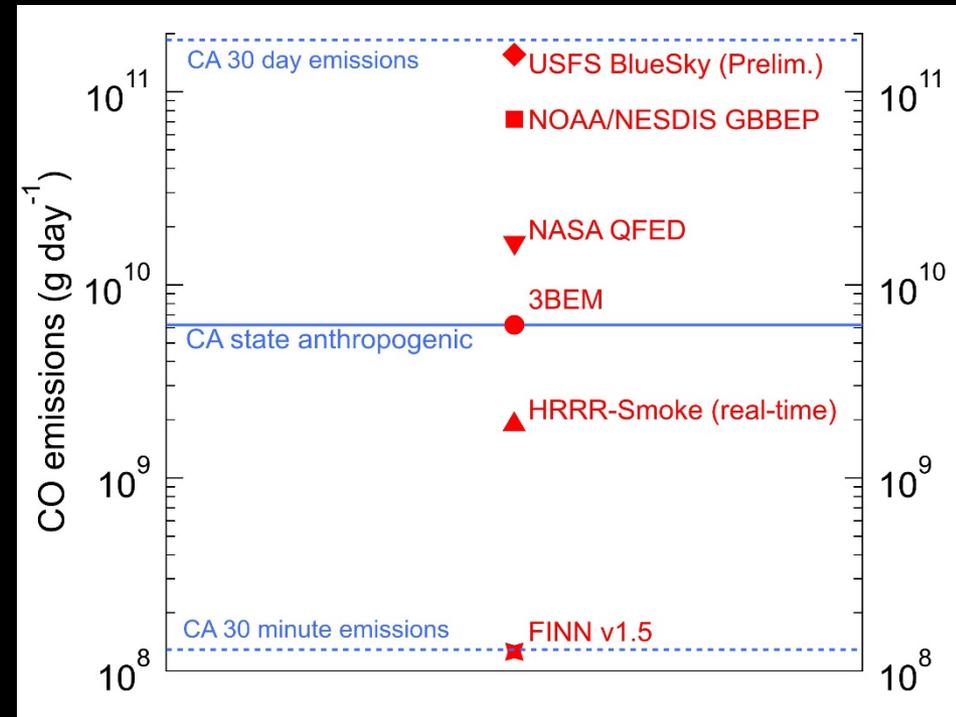
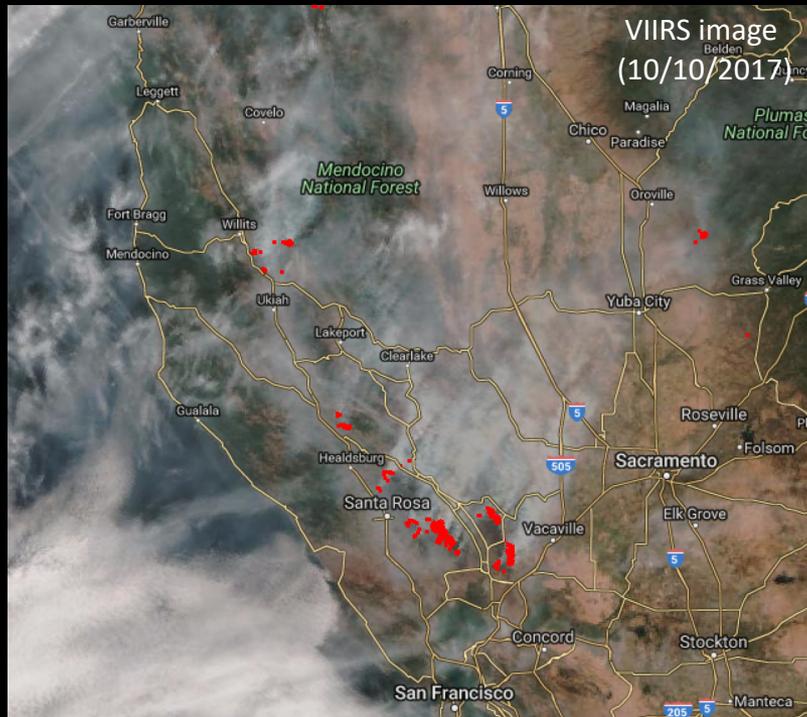
Kurtis Alexander

Nov. 30, 2018

Updated: Nov. 30, 2018 4 a.m.

Analytical challenge: lack of measurement techniques to evaluate emissions from wildfires.

October 2017 Northern California fires

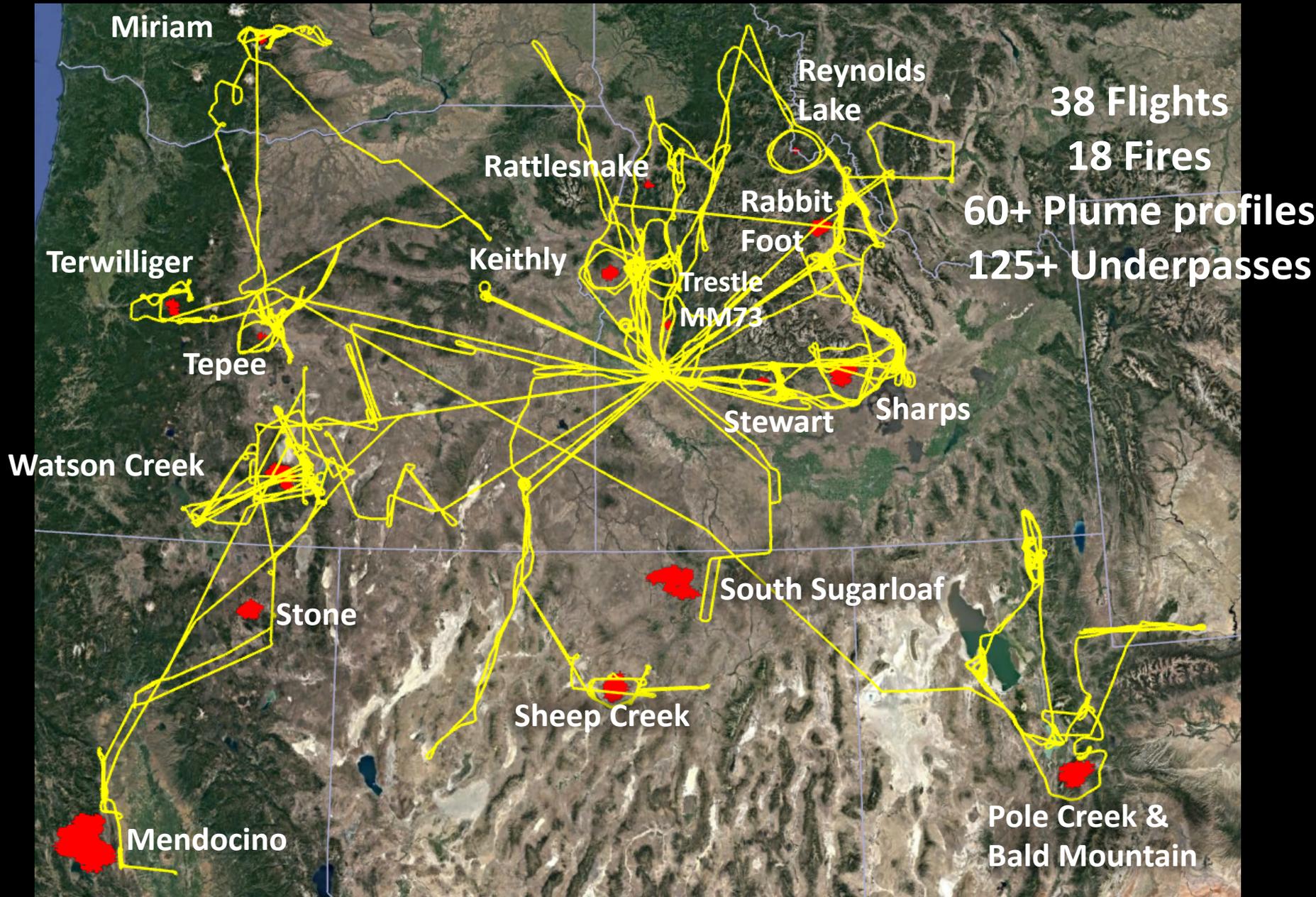


Emissions estimates for Santa Rosa fires vary by >3 orders of magnitude!

Variability is driven by uncertainty in amount of fuel burned.

Hypothesis: It is possible to quantify trace gas fluxes at the scale of wildfires.

BB-FLUX 2018



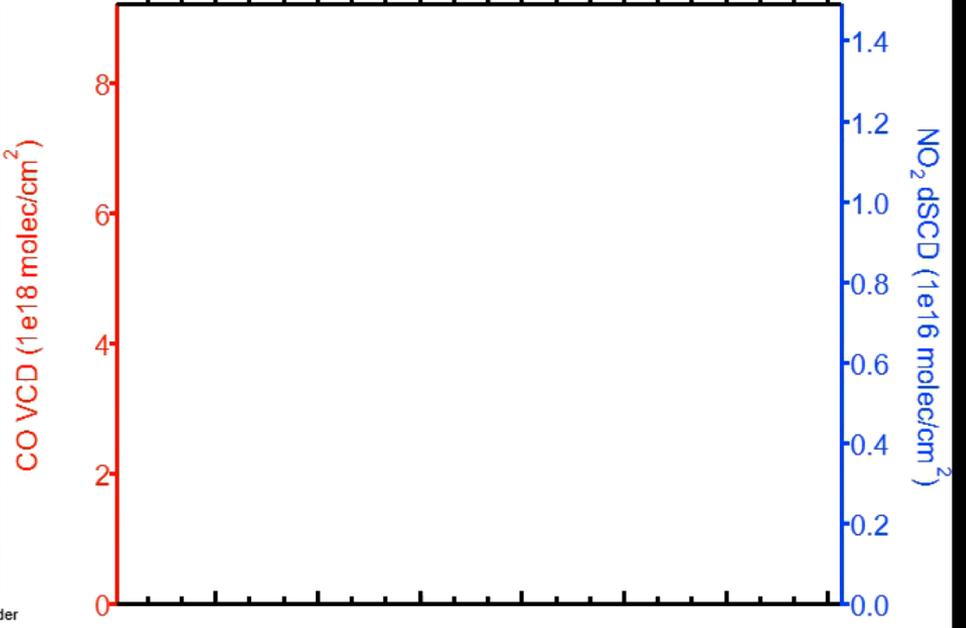
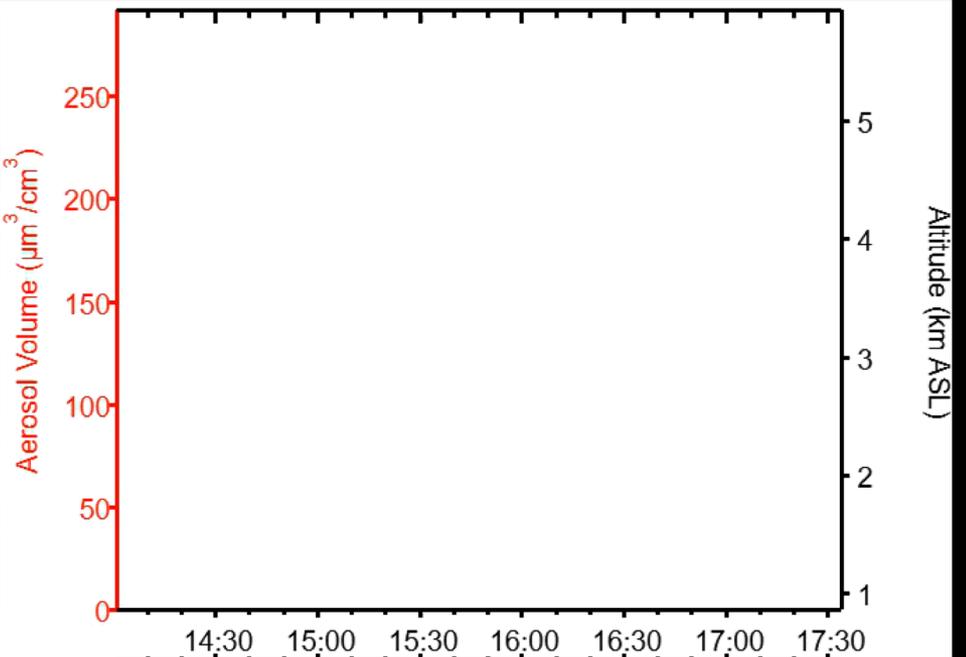
Rabbit Foot Fire, ID



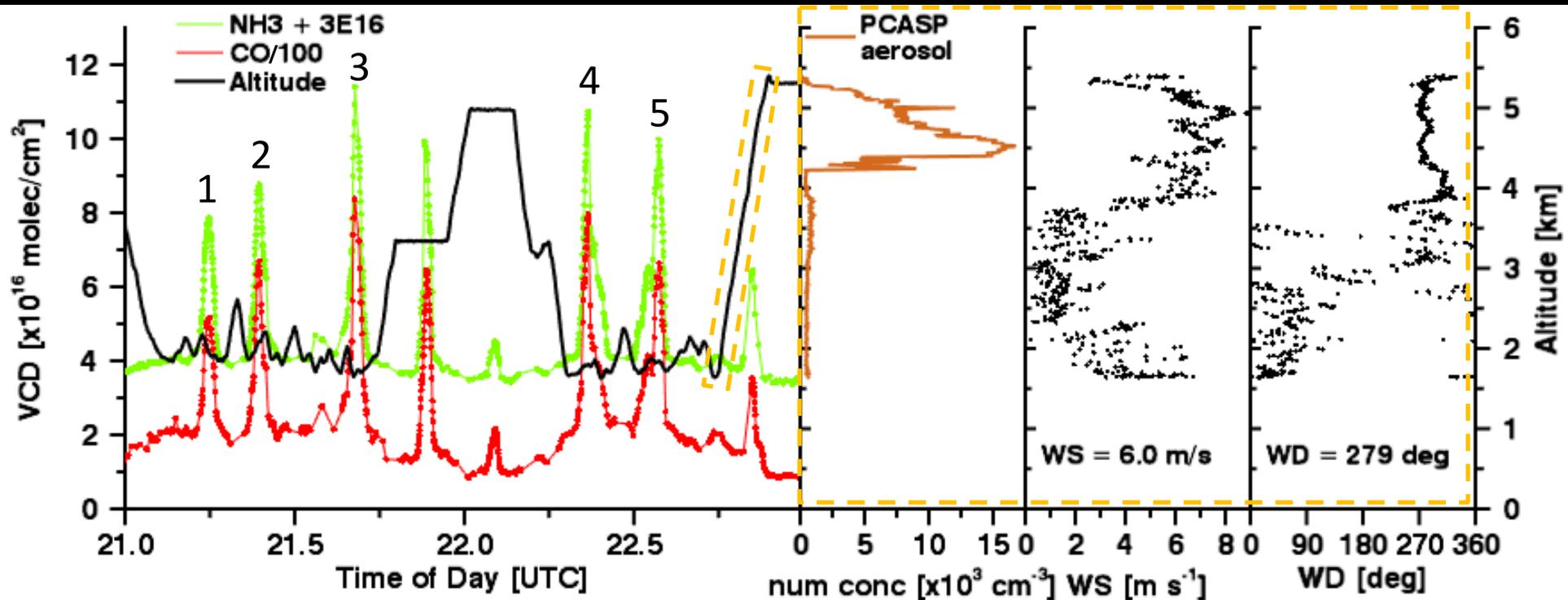
08 Aug 2018 14:01 Pitch: ° Roll: ° T: °C RH: %



GOES-16 images from NASA/NOAA, processed by L. Oolman, U. Wyom., Video by K. Zarzana, Volkamer Group, CU Boulder



Emission Fluxes from CU SOF



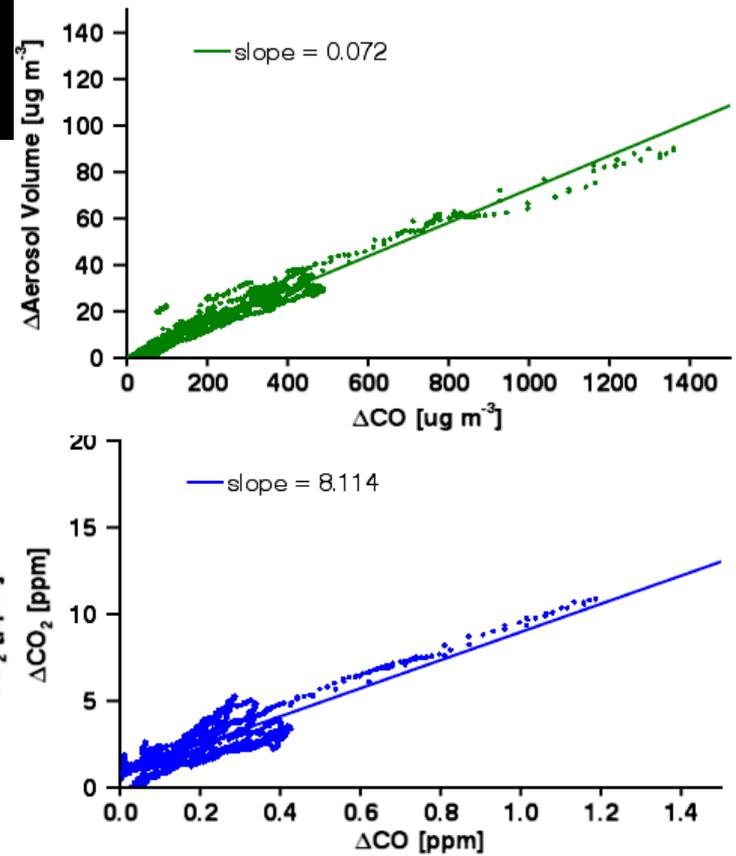
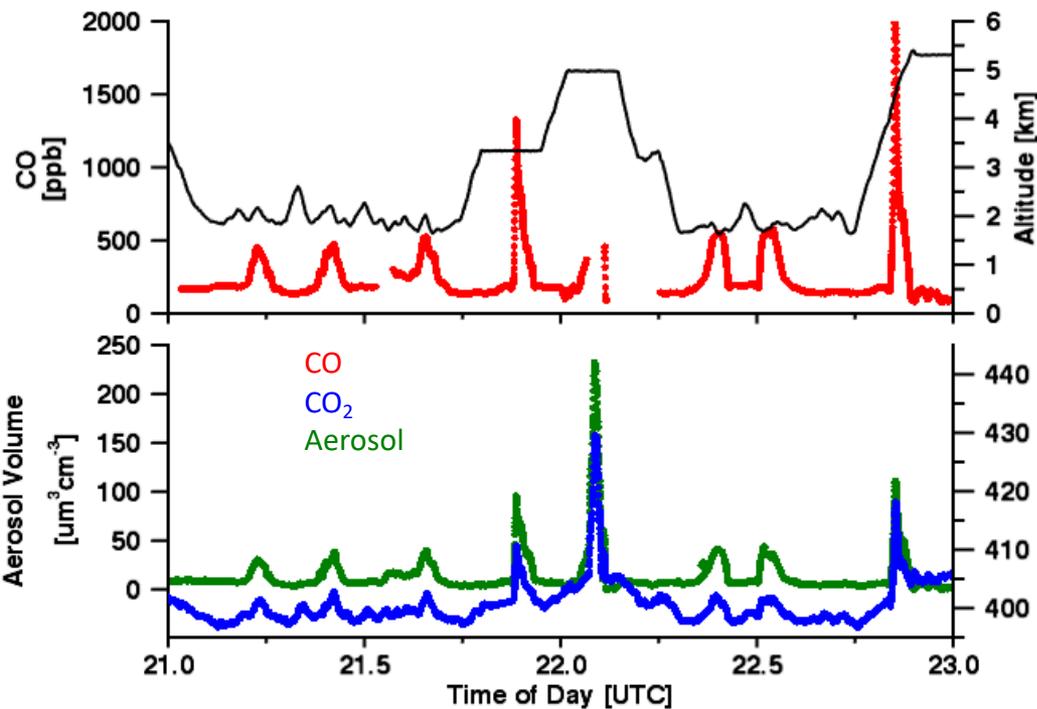
$$Flux = \int_S \Delta VCD \vec{n} \cdot \vec{F} ds$$

$$\Delta VCD = VCD - VCD_{BKG}$$

$$\vec{F} = \text{wind} \quad \vec{n} = \text{normal to flight direction}$$

	1	2	3	4	5
CO [t/hr]	382 - 409	476 - 508	663 - 709	749 - 800	622 - 665
NH ₃ [t/hr]	2.8 - 3.0	3.3 - 3.6	4.4 - 4.8	4.8 - 5.1	4.2 - 4.5

Total Carbon Fluxes

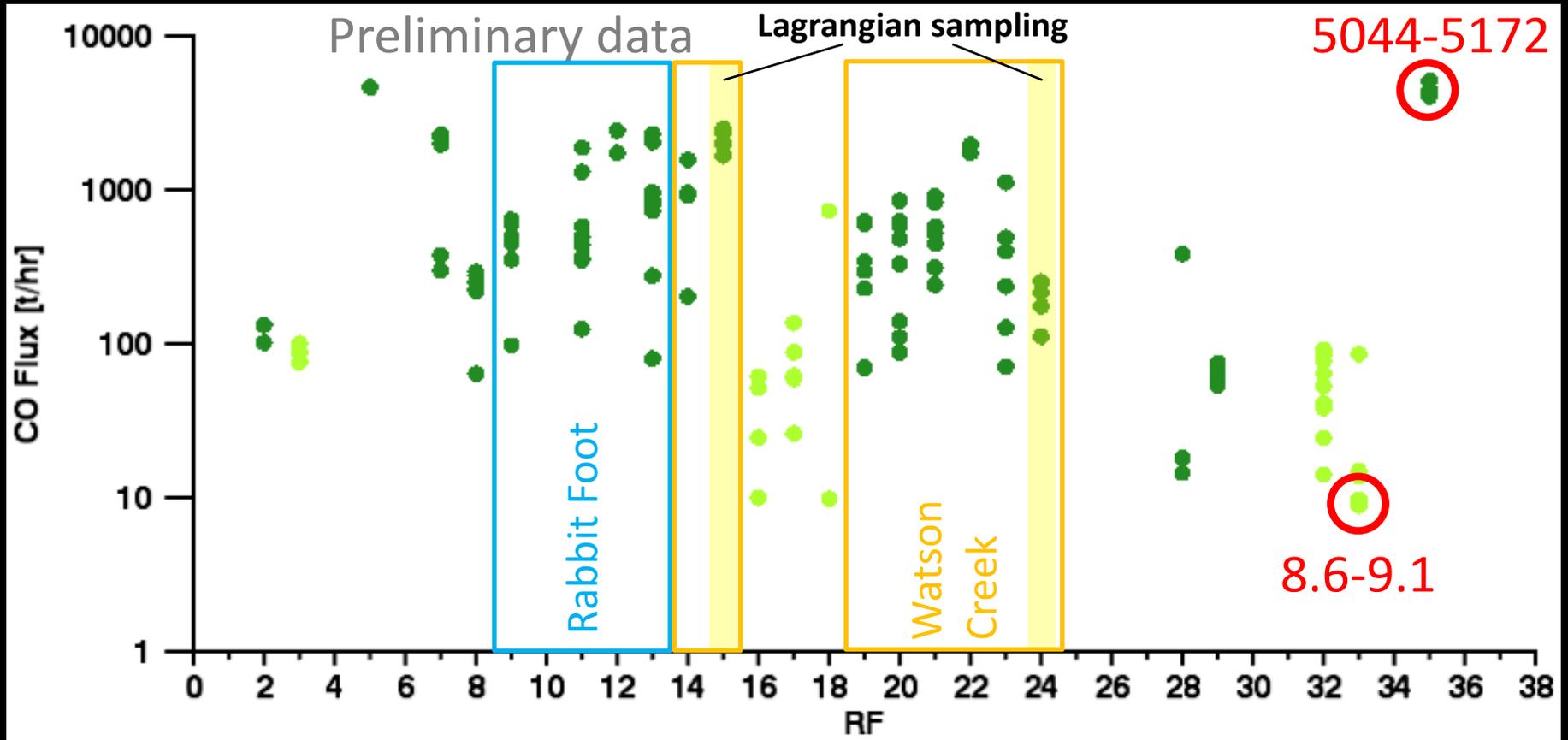


$$Flux(Z) = ER_{w/w}(Z) \cdot CO\ Flux$$

$$ER_{w/w}(Z) = \frac{\Delta Z}{\Delta CO} \quad Z = CO_2, \dots$$

	1	2	3	4	5
CO [t/hr]	382 - 409	476 - 508	663 - 709	749 - 800	622 - 665
CO ₂ [t/hr]	3100 - 3319	3862 - 4122	5380 - 5753	6077 - 6491	5047 - 5396
Aerosol ¹ [t/hr]	28 - 29	34 - 37	48 - 51	54 - 58	45 - 48

SOF perspective of BB-FLUX

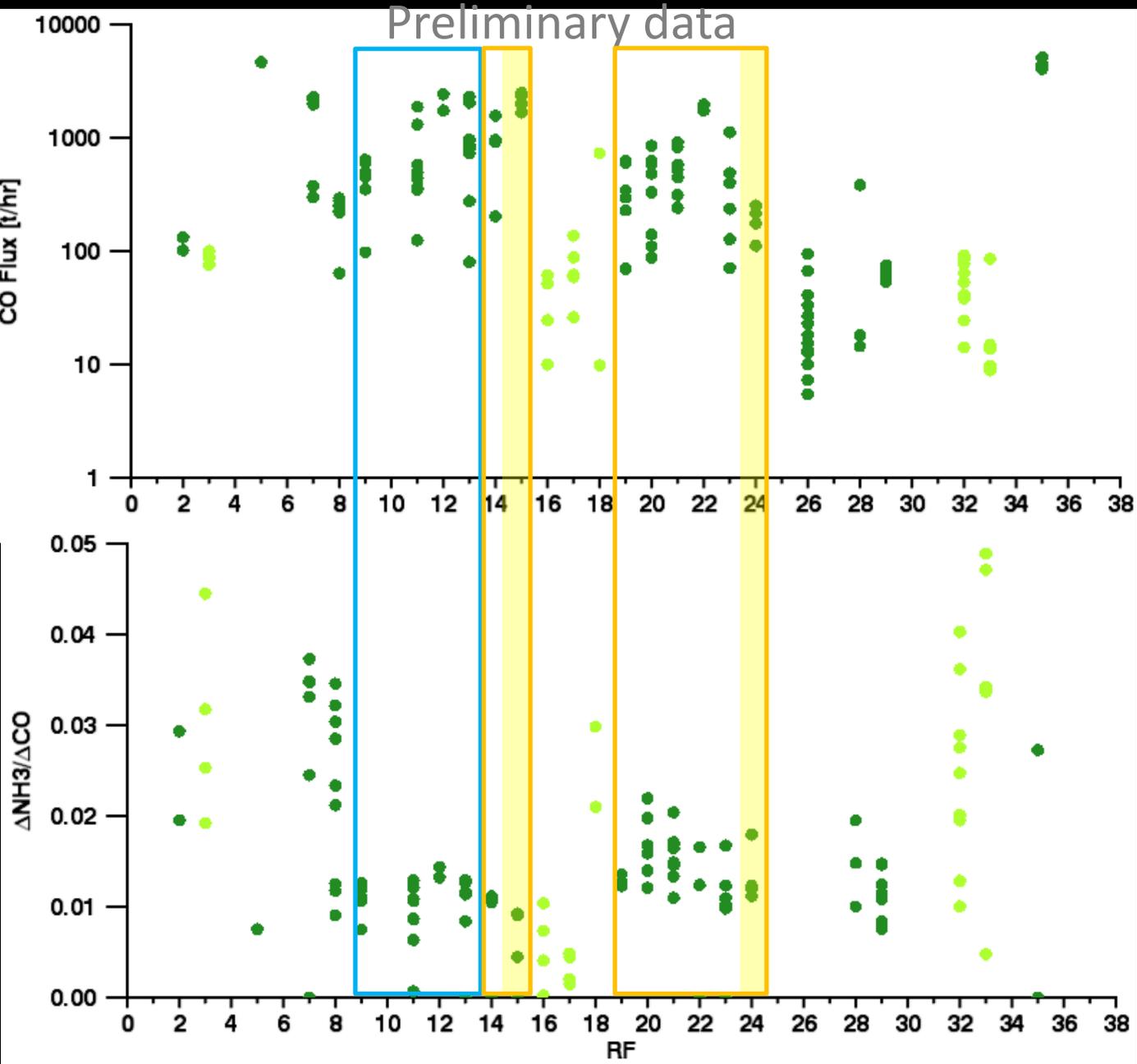


Primary fuel type

- grass
- tree

SOF captures fluxes in large dynamic range

SOF perspective of BB-FLUX



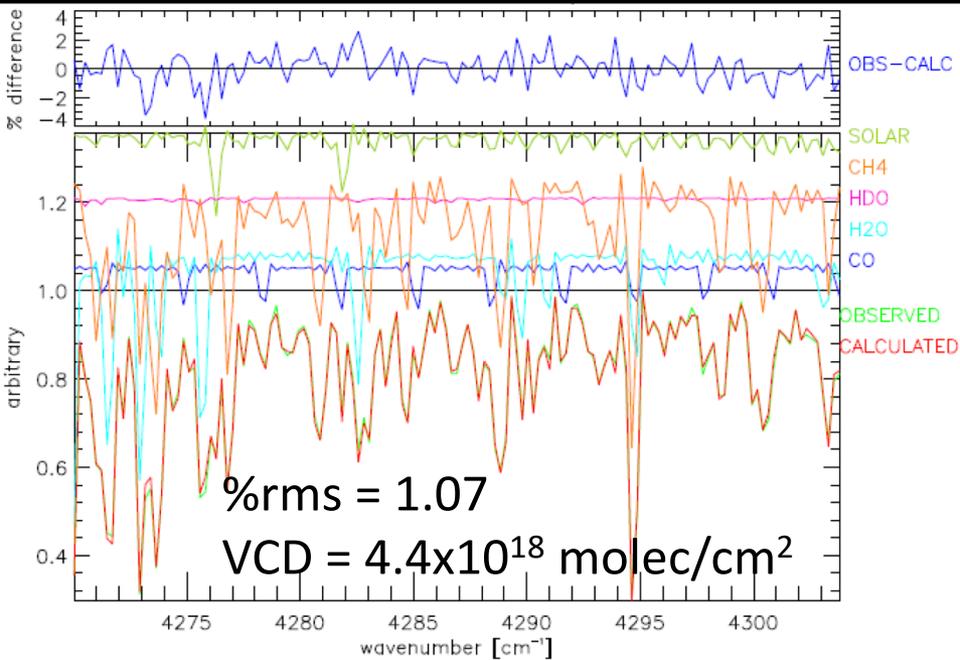
Conclusions

- Airborne SOF is a **new tool to quantitatively evaluate fire emissions**
 - Measured **CO fluxes from 8.6 to 5200 t/hr**
- Synergistic application of SOF and ERs (in situ or column) is powerful to quantify and speciate **trace gas and aerosol fluxes**
 - **Total carbon flux** dominated by CO₂ and CO, smaller contributions from aerosol and VOCs
- SOF observed **$\Delta\text{NH}_3/\Delta\text{CO} < 0.01$ to 0.05**
- **Outlook:**
 - **Develop new SOF retrievals**
 - **Characterize time evolution of emission fluxes and ratios**

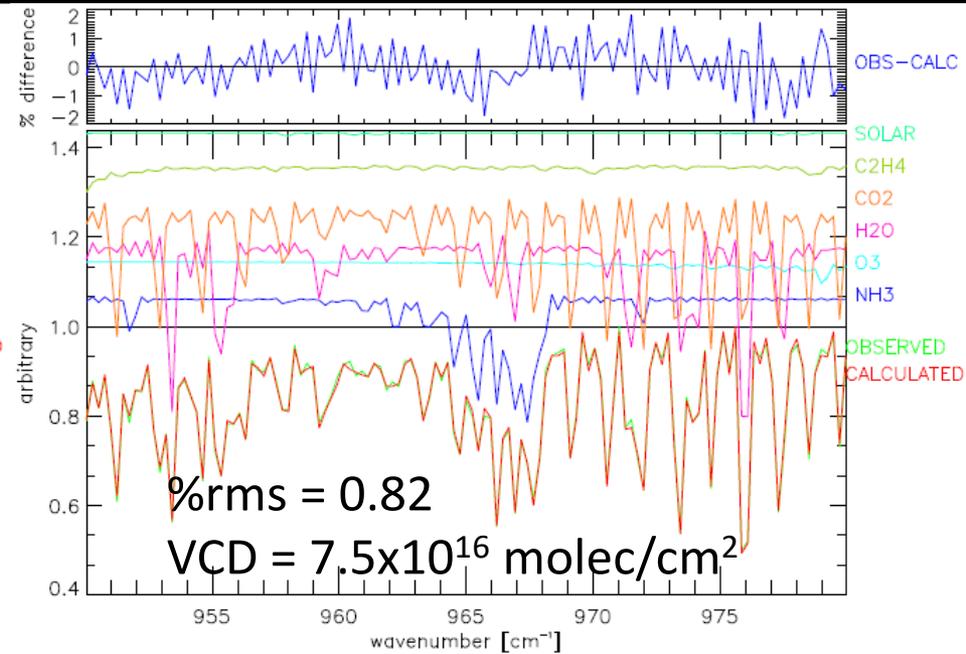


Retrieval windows

CO

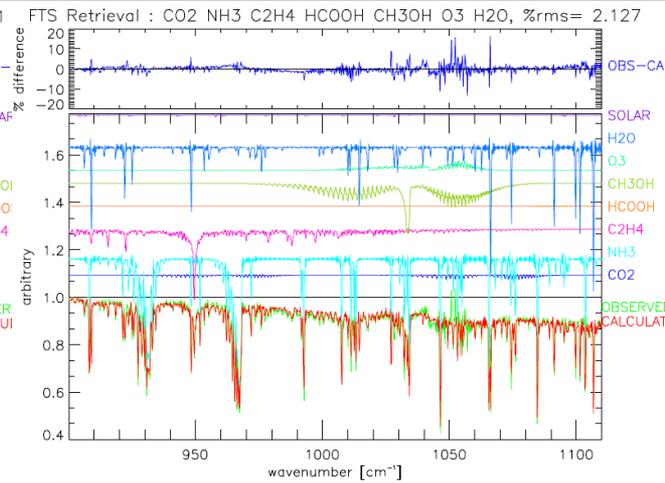
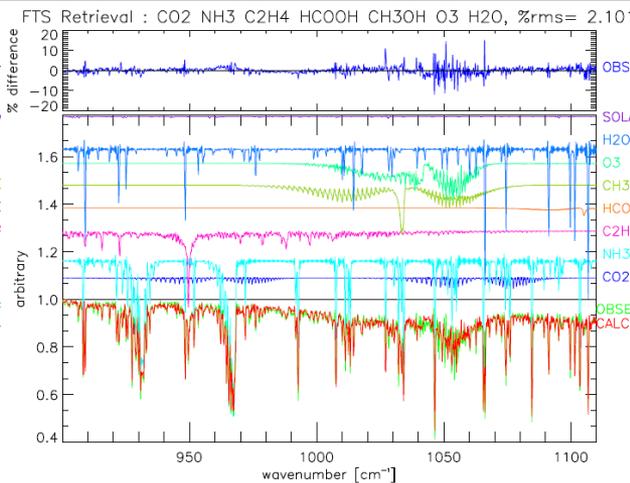
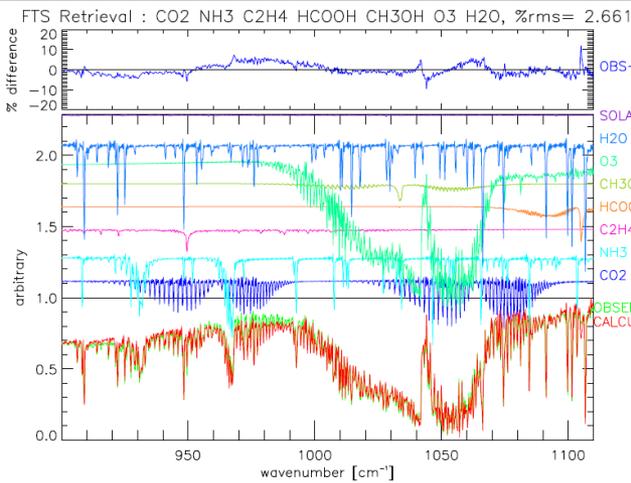


NH₃



2sec in plume spectrum

(2sec in plume)/(2sec outside) (16sec in plume)/(16sec outside)



Running retrieval as usual.

Comparison:

	2 se spec	Ratio	Avg, ratio
Iterations	7	25	28
CO2 [$\times 10^{21}$ molec/cm ²]	5.7	0.5	0.2
O3 [$\times 10^{18}$ molec/cm ²]	9.0	0.1	Negative (-0.3)
H2O [$\times 10^{22}$ molec/cm ²]	2.0	0.6	0.6
CH3OH [$\times 10^{16}$ molec/cm ²]	6.2	11.2	11.7