Estimating Biomass Burning Emissions with Toronto FTIR Data

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NDACC IRWG & TCCON Meeting 23 May 2019 Wanaka, New Zealand



U of Toronto Atmospheric Observatory (TAO)



Location: 43.66N, 79.40W, 174 masl

Solar tracker



- Primary instrument:
 - ABB Bomem DA8 Fourier transform infrared (FTIR) spectrometer
- Measurements started in 2002
- Urban site (downtown Toronto)

Scientific Objectives

- To assess biomass burning emissions and quantify their emission factors
 - Use TAO FTIR measurements to identify enhancements due transport of biomass burning emissions
 - Use models and satellite data for source attribution
- Other scientific objectives include:
 - To investigate the daily, seasonal, and interannual trace gas variability and trends in an urban setting (downtown Toronto)
 - To characterize the origin of urban pollution events (local or long-range transport) with models (e.g., the GEOS-Chem chemical transport model)

Biomass Burning

- Biomass burning emissions can negatively affect air quality
- Emissions can have photochemical and radiative forcing effects, particularly when plumes are transported to Arctic regions (Amiro et al., 2001)
- Quantifying biomass burning emissions and understanding their transport poses a challenge
 - Types of vegetation burned, the combustion phase (smoldering vs. flaming), and atmospheric conditions at the time of the fire must all be accounted for when quantifying emissions
 - Emitted gases may also decay and undergo chemistry in the atmosphere during transport

Biomass Burning Enhancement Events

- Wildfires emit numerous chemical species including CO, C₂H₆, HCN, HCOOH, H₂CO, CH₃OH and others
- Given their long lifetimes, CO, HCN and C₂H₆ are good tracers of biomass burning events (e.g., Viatte et al., 2013)
- Formic acid, methanol and formaldehyde have shorter lifetimes

• Typical lifetimes (τ) are: 61 days (CO), 45 days (C₂H₆), ~150 days (HCN), 4 days (HCOOH), 7.5 days (CH₃OH) and ~1 day (H₂CO)

(Viatte et al., 2013, Millet et al., 2015, Jacob et al., 2005, Pommier et al., 2017)

Biomass Burning Analysis

- Enhancements identified in August 2012, July 2015 and September 2017
 - Simultaneous enhancements of CO, HCN, C₂H₆
 - For 2015 and 2017, HCOOH and CH₃OH enhancements were also seen
- FLEXPART used for source attribution
- Travel-time estimated with HYSPLIT
 - Travel-time used to estimate decay in the atmosphere
- Emission factors and emission ratios reported
- 2015 event analysis will be presented in detail here

Biomass Burning Analysis



TAO timeseries for CO, HCN, C_2H_6 , CH₃OH and HCOOH

- The arrows indicate the 2012, 2015 and 2017 events (in CO observations)
- Also of note is the clear enhancement in 2014 (Lutsch et al., 2016)

2015 Early July

- Sharp peak in HCN on July 3rd to 6th (indicated by arrows)
- Gray band indicates 2 standard deviations above and below the monthly mean





Source Attribution with FLEXPART: 2015

FLEXPART sensitivity plot for July 3rd

Total Column Sensitivity: AIRTRACER

MODIS fire product data for June 29th



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Travel Time and Decay

 Chemical species decay exponentially in the atmosphere

• i.e.
$$[x]_t = [x]_0 e^{-t/\tau}$$

where τ denotes the lifetime

 HYSPLIT back trajectory was ran with reanalysis meteorological field for travel time estimation

NOAA HYSPLIT MODEL Backward trajectories ending at 1400 UTC 03 Jul 15 CDC1 Meteorological Data



Tagged CO GEOS-Chem (v12)

GEOS-Chem tagged CO



GEOS-Chem surface CO for 2015 July event

2015-06-20 00:00:00 GEOS-Chem CO



Not shown here are similar surface GEOS-Chem outputs for other biomass burning events



Tagged CO Full Timeseries



Note the red spike in 2015

FTIR and GEOS-Chem comparison (top) Relative contribution (bottom)

Interpreting the Data

- Enhancement Ratio (EnhR)
 - Ratio of column of species of interest to column of CO
 - Calculated for HCN, C₂H₆, CH₃OH and HCOOH
 - EnhR is the slope of the linear regression when the two species are plotted against each other
 - To account for the travel time decay and to calculate the emission ratio (ER), the following equation is used:

$$ER_{X} = EnhR_{X}\left(\frac{e^{t/\tau_{X}}}{e^{t/\tau_{CO}}}\right)$$

ER and EF

- ER: Emission Ratio
 - Enhancement ratio with plume aging correction
 - Since travel times are accounted for, it is not location specific
- EF: Emission Factor

Ο

Defined by:

$$EF_X = EF_{CO} \cdot ER_X \cdot \left(\frac{MW_X}{MW_{CO}}\right)$$

 where EF_{CO} is the emission factor of CO, which is 127 ± 45 g/kg (Akagi et al., 2011), and MW are the molecular weights

Not location specific

EnhR_x Plots



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Results: HCN

Source	Platform	$EF_{co}~(g/Kg)$	ER	EF (g/Kg)
			HCN	
^[a] 2012 August	FTIR	89 ± 32	0.0038 ± 0.0006	0.33 ± 0.13
^[b] 2012 August	FTIR	89 ± 32	0.0038 ± 0.0006	0.32 ± 0.13
^[a] 2015 July	FTIR	127 ± 45	0.0040 ± 0.0002	0.49 ± 0.18
^[b] 2015 July	FTIR	127 ± 45	0.0039 ± 0.0002	0.48 ± 0.17
^[a] 2017 September	FTIR	127 ± 45	0.0038 ± 0.0003	0.47 ± 0.17
^[b] 2017 September	FTIR	127 ± 45	0.0037 ± 0.00027	0.45 ± 0.16
^[a] 2017 September	FTIR	89 ± 32	0.0038 ± 0.00027	0.33 ± 0.12
^[b] 2017 September	FTIR	89 ± 32	0.0037 ± 0.00027	0.32 ± 0.12
^[e] Lutsch et al. (2016)	FTIR	127 ± 45	0.0068 ± 0.0003	0.84 ± 0.30
^[c] Viatte et al. (2015)	FTIR	127 ± 45	0.00429 ± 0.00245	0.44 ± 0.25
^[d] Viatte et al. (2015)	FTIR	127 ± 45	0.00343 ± 0.00094	0.36 ± 0.17
^[f] Paton-Walsh et al. (2005)	FTIR	107 ± 37	0.0043 ± 0.0016	0.43 ± 0.22
Akagi et al. (2011)	Compilation	127 ± 45	-	1.52 ± 0.82
Akagi et al. (2011)	Compilation	89 ± 32	-	0.73 ± 0.19
$^{[g]}$ Goode et al. (2000)	Aircraft	88.8	0.0069	0.61

^{[a],[b]}This study.

[a]HCN and CO lifetimes 150 and 61 days respectively.
[b]HCN and CO lifetimes 75 and 30 days respectively.
[c],[d]Values from Thule and Eureka respectively.
[e]Values from Toronto.

^[f]Values from Australian fires.

^[g]Values from Alaskan fires.

Results: C₂H₆

Source	Platform	EF_{co} (g/Kg)	ER	EF (g/Kg)
			C_2H_6	
^[a] 2012 August	FTIR	89 ± 32	0.0091 ± 0.0012	0.87 ± 0.33
^[b] 2012 August	FTIR	89 ± 32	0.0089 ± 0.0012	0.85 ± 0.33
^[a] 2015 July	FTIR	127 ± 45	0.013 ± 0.0006	1.79 ± 0.64
^[b] 2015 July	FTIR	127 ± 45	0.013 ± 0.0006	1.72 ± 0.62
^[a] 2017 September	FTIR	127 ± 45	0.019 ± 0.0011	2.64 ± 0.95
^[b] 2017 September	FTIR	127 ± 45	0.019 ± 0.0011	2.53 ± 0.91
^[a] 2017 September	FTIR	89 ± 32	0.019 ± 0.0011	1.85 ± 0.67
^[b] 2017 September	FTIR	89 ± 32	0.019 ± 0.0011	1.77 ± 0.65
^[e] Lutsch et al. (2016)	FTIR	127 ± 45	0.0101 ± 0.0005	1.38 ± 0.49
^[c] Viatte et al. (2015)	FTIR	127 ± 45	0.01211 ± 0.00476	1.39 ± 0.68
^[d] Viatte et al. (2015)	FTIR	127 ± 45	0.00956 ± 0.00319	1.09 ± 0.74
$^{[f]}$ Paton-Walsh et al. (2005)	FTIR	107 ± 37	0.0023 ± 0.0005	0.26 ± 0.11
Akagi et al. (2011)	Compilation	127 ± 45	-	1.79 ± 1.14
Akagi et al. (2011)	Compilation	89 ± 32	-	1.12 ± 0.67
$^{[g]}$ Goode et al. (2000)	Aircraft	88.8	0.0073	0.66

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^[a]HCN and CO lifetimes 150 and 61 days respectively.
^[b]HCN and CO lifetimes 75 and 30 days respectively.
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^[g]Values from Alaskan fires.

Results: CH₃OH

Source	Platform	EF_{co} (g/Kg)	ER	EF (g/Kg)	
			CH ₃ OH		
^[a] 2012 August	FTIR	89 ± 32	-	-	
^[b] 2012 August	FTIR	89 ± 32	-	-	
^[a] 2015 July	FTIR	127 ± 45	0.030 ± 0.0064	4.35 ± 1.80	
^[b] 2015 July	FTIR	127 ± 45	0.029 ± 0.0056	4.17 ± 1.68	
^[a] 2017 September	FTIR	127 ± 45	0.045 ± 0.008	6.54 ± 2.60	
^[b] 2017 September	FTIR	127 ± 45	0.043 ± 0.007	6.27 ± 2.43	
^[a] 2017 September	FTIR	89 ± 32	0.045 ± 0.008	4.58 ± 1.85	
^[b] 2017 September	FTIR	89 ± 32	0.043 ± 0.007	5.39 ± 1.72	
^[e] Lutsch et al. (2016)	FTIR	127 ± 45	-	-	
^[c] Viatte et al. (2015)	FTIR	127 ± 45	-	-	
^[d] Viatte et al. (2015)	FTIR	127 ± 45	0.02813 ± 0.01252	3.44 ± 1.68	
^[f] Paton-Walsh et al. (2005)	FTIR	107 ± 37	-	-	
Akagi et al. (2011)	Compilation	127 ± 45	-	2.82 ± 1.62	
Akagi et al. (2011)	Compilation	89 ± 32	-	1.93 ± 1.38	
$^{[g]}$ Goode et al. (2000)	Aircraft	88.8	0.0135	1.41	

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^[a]HCN and CO lifetimes 150 and 61 days respectively. ^[b]HCN and CO lifetimes 75 and 30 days respectively.

^{[c],[d]}Values from Thule and Eureka respectively.

^[e]Values from Toronto.

^[f]Values from Australian fires.

^[g]Values from Alaskan fires.

Results: HCOOH

Source	Platform	EF_{co} (g/Kg)	ER	EF (g/Kg)	
			HCOOH		
^[a] 2012 August	FTIR	89 ± 32	-	-	
^[b] 2012 August	FTIR	89 ± 32	-	-	
^[a] 2015 July	FTIR	127 ± 45	0.015 ± 0.0012	3.07 ± 1.12	
^[b] 2015 July	FTIR	127 ± 45	0.014 ± 0.0012	2.94 ± 1.07	
^[a] 2017 September	FTIR	127 ± 45	0.0071 ± 0.0034	1.48 ± 0.88	
^[b] 2017 September	FTIR	127 ± 45	0.013 ± 0.008	2.64 ± 1.87	
^[a] 2017 September	FTIR	89 ± 32	0.0071 ± 0.0034	1.03 ± 0.62	
^[b] 2017 September	FTIR	89 ± 32	0.013 ± 0.008	1.85 ± 1.31	
^[e] Lutsch et al. (2016)	FTIR	127 ± 45	-	-	
^[c] Viatte et al. (2015)	FTIR	127 ± 45	$0.01790\ {\pm}0.00937$	3.14 ± 1.46	
^[d] Viatte et al. (2015)	FTIR	127 ± 45	0.01531 ± 0.00403	2.69 ± 1.14	
^[f] Paton-Walsh et al. (2005)	FTIR	107 ± 37	-	-	
Akagi et al. (2011)	Compilation	127 ± 45	-	0.57 ± 0.46	
Akagi et al. (2011)	Compilation	89 ± 32	-	0.35 ± 0.33	
$^{[g]}$ Goode et al. (2000)	Aircraft	88.8	0.0062	0.99	

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^[a]HCN and CO lifetimes 150 and 61 days respectively.

^[b]HCN and CO lifetimes 75 and 30 days respectively.

^{[c],[d]}Values from Thule and Eureka respectively.

^[e]Values from Toronto.

^[f]Values from Australian fires.

^[g]Values from Alaskan fires.

Data Analysis

• Trend analysis

Fitting a trended Fourier series

- Pollution events
 - Pollution events can be identified by using residuals of the fit (Zellweger et al., 2009)
- Statistical significance of trends can be examined using a method outlined by Weatherhead et al. (1998)
- Bootstrap resampling to find confidence intervals (Gardiner et al., 2008)

Long-term Trend Analysis

- Trend analysis:
 - Fitting a trended Fourier series:

$$F(t) = a_0 + \frac{a \cdot t}{T} + \sum_{\omega \in \Omega} \alpha_\omega \sin\left(\frac{\omega t}{T}\right) + \beta_\omega \cos\left(\frac{\omega t}{T}\right)$$

where T is the total time of the data

- Pollution events are defined by:
 - Finding residuals, ie. Observation minus F(t)
 - Mirroring negative residuals as well as calculating the standard deviation of the residuals
 - Any measurements that are 2 standard deviations or higher than the fit and residuals are considered pollution events

Trend Analysis and Pollution Event Identification

- For both trend analysis and identifying pollution events (by looking at residuals), fitting a trended Fourier series is needed
- How do we know if we are fitting the data correctly?
 - Try several fits of different orders and varying frequencies
 - Look for correlation between the fit and the data
 - Look at the residuals and test for normality (using Kolmogorov-Smirnov test)
 - If the fit captures all of the natural variability and trend in the data, then the residuals will tend towards a normal distribution given enough sample size due to central limit theorem
 - If residuals exhibit normality at low order but not in higher orders, it means an artificial pattern was added to the data i.e. overfitting

Fitting TAO Data: *Preliminary Results*

- Even at high orders (6+) over fitting was not observed
- Higher order fits led to marginally better correlation
 and lower RMS in the residuals
- Higher order fits did not change the number of years needed for trend detection
 - \circ E.g. CH₄: ~12 years, N₂O: ~17 years, C₂H₆: ~26 years
- Currently working on
 - Utilizing other tests of normality (Lilliefors and Jarque–Bera)
 - Allowing the amplitude of the annual cycle to change over time (linearly)
 - Effects of varying the Q-value (for bootstrap reanalysis)
 - Applying similar analyses to other NDACC data

Summary and Discussion

- The 15+ years of data collected at TAO were analyzed:
 - Analysis of several biomass burning events
 - Examination of trends and pollution events
- Emission factors of several biomass burning events seen in Toronto were calculated
 - FLEXPART in conjunction with MODIS Fire Product was used for source attribution
 - GEOS-Chem tagged CO simulation was used to supplement the FTIR observations
- Statistics for trend analysis methods were examined in the context of TAO data

Future Work

- Trend analysis of hydrocarbons and pollutants
- Analysis of pollution events with a focus on $\rm O_3$ and CO
 - $\circ~$ Utilize tagged O_3 GEOS-Chem model for in-depth analysis of Toronto O_3 ~
- Further integration of measurements and modeling (GEOS-Chem)
- AmmonAQ project in collaboration with LATMOS
 Analysis of urban air quality focusing on NH₃, using
 - Toronto and Paris as benchmark cities
- Comparison of urban (TAO) and rural (ECCC Egbert) FTIR measurements
 - Egbert is about 80 km north of downtown Toronto (TAO)
 - Egbert FTIR is in good working condition, a new sun tracker has been installed

Egbert FTIR & Sun Tracker

Bruker solar tracker installed here





DA8 FTIR→

sunlight



Egbert FTIR & Sun Tracker

- Egbert can complement TAO data
- As a rural site, it can be a point of comparison for an urban site like TAO
- High-resolution models (GEOS-Chem) can differentiate between the two sites





Acknowledgements

- NSERC CREATE Training Program in Technologies for Exo-Planetary Science (TEPS)
- ABB Bomem, CFCAS, CFI, CRESTech, CSA, ECCC, NSERC, ORDCF, PREA, University of Toronto
- The many interns, students, and postdocs who have made TAO FTIR measurements, including (most recently): Orfeo Colebatch, Ellen Eckert, Lei Liu, Erik Lutsch, and Tyler Wizenberg



Additional Slides: Bootstrap Reanalysis

- Bootstrap resampling is done to find confidence intervals
 - It is an analysis where residuals are randomly redistributed to form a "new" set of data, where another line is refitted
 - This process is repeated (over several hundred to thousands of times)
 - $_{\odot}$ The ensemble of data is analyzed to find the 2- σ confidence interval (2.5% to 97.5% coverage of the data)

Additional Slides: Weatherhead Method

- Detection of long-term, linear trends is affected by a number of factors
 - Size of the trend to be detected
 - Time span of the data
 - Magnitude of variability and autocorrelation
- The number of years of data needed to detect a trend strongly depends on (and increases with) the magnitude of variance and the autocorrelation coefficient
- Environmental time series data are often autocorrelated

Additional Slides: More on Biomass Burning

 "As burning occurs, it can release hundreds of years worth of stored carbon dioxide into the atmosphere in a matter of hours"

NASA, Earth Observatory (earthobservatory.nasa.gov)

 "Exposure to biomass burning particles is strongly associated with cardiovascular disease, respiratory illness, lung cancer, asthma and low birth weights"

Stanford News, 2014 (news.stanford.edu)

Additional Slides: Atmospheric Formic Acid

- Formic acid is one of the most abundant acids in the atmosphere
- Greatly contributes to free precipitation acidity
- Affects aqueous phase chemistry

 pH-dependent reaction rates
- Naturally produced photochemically, though emission sources (including anthropogenic) also certainly exist

Additional Slides: Retrieval

- A retrieval algorithm called SFIT4 is used to derive vertical profiles and/or columns of trace gases from their absorption and emission spectra:
 - Identify spectral lines of interest
 - Generate a model atmosphere using meteorology data from NCEP and WACCM
 - Use a forward model to simulate a model spectrum
 - Iteratively adjust the *a priori* VMR profile until the model spectrum agrees with the measured spectrum
- SFIT4 is an optimal estimation method (OEM) analysis
 - It uses both measurement data and the *a priori* information
 - It assigns weights to the *a priori* information and the measurement based on each of their uncertainties